

DIFFUSION: MICROSCOPIC AND MACROSCOPIC THEORY

PROF. ARAVI SAMUEL

DAVID ZIMMERMAN

PHYSICS/NEURO 141

WEEK 5



MICROSCOPIC THEORY

BROWNIAN MOTION



Lysozyme

- Mass = $m = 2.3 \times 10^{-20} \text{g}$
- $k_B T = 4 \times 10^{-14} \text{g cm}^2 / \text{sec}^2$
- $\langle v_x^2 \rangle^{1/2} = 1.3 \times 10^3 \text{cm/sec}$



1D random walk

- Particles start at the origin
- Step Size = δ
- Step Interval = τ
- Instantaneous velocity = δ/τ

BROWNIAN MOTION, THE FIRST MOMENT

- Each particle steps to the right or to the left once every τ seconds, moving at velocity $\pm v_x$ a distance $\delta = \pm v_x \tau$.
- the position of a particle after the n th step differs from its position after the $(n - 1)$ th step by $\pm \delta$:

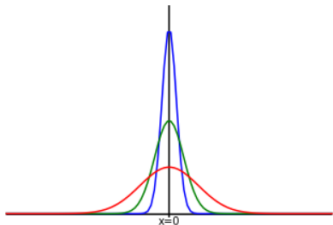
$$x_i(n) = x_i(n - 1) \pm \delta$$

- The mean displacement does not change:

$$\begin{aligned}\langle x(n) \rangle &= \frac{1}{N} \sum_{i=1}^N [x_i(n - 1) \pm \delta] \\ &= \frac{1}{N} \sum_{i=1}^N x_i(n - 1) \\ &= \langle x(n - 1) \rangle\end{aligned}$$

- If $\langle x(0) \rangle = 0$, then $\langle x(t) \rangle = 0$

BROWNIAN MOTION, THE SECOND MOMENT



- The probability of finding particles at different points at $t = 1, 4,$ and 16 sec.
- Standard deviations increase with the square root of time

Squared coordinate of each particle changes with time:

$$x_i^2(n) = x_i^2(n-1) \pm 2\delta x_i(n-1) + \delta^2$$

Mean squared coordinates of all particles grows with time:

$$\begin{aligned}\langle x^2(n) \rangle &= \frac{1}{N} \sum_{i=1}^N x_i^2(n) \\ &= \frac{1}{N} \sum_{i=1}^N [x_i^2(n-1) \pm 2\delta x_i(n-1) + \delta^2] \\ &= \langle x^2(n-1) \rangle + \delta^2\end{aligned}$$

THE DIFFUSION COEFFICIENT

- Since $x_i(0) = 0$ for all particles i , $\langle x^2(0) \rangle = 0$. Thus, $\langle x^2(1) \rangle = \delta^2$, $\langle x^2(2) \rangle = 2\delta^2$, $\langle x^2(3) \rangle = 3\delta^2$, ..., and $\langle x^2(n) \rangle = n\delta^2$.
- It follows that the mean-square displacement is proportional to t , the root-mean-square displacement is proportional to the square-root of t . Note that $n = t/\tau$, so that:

$$\langle x^2(t) \rangle = (t/\tau)\delta^2 = (\delta^2/\tau)t$$

- For convenience, we define a diffusion coefficient, $D = \delta^2/2\tau$. This gives us

$$\langle x^2 \rangle = 2Dt$$

THE DIFFUSION COEFFICIENT

DRAW ON A SPHERE

The Equations for Fluid Dynamics

■ Navier-Stokes Equations

$$-\nabla p + \eta \nabla^2 \mathbf{v} = \rho \frac{\partial \mathbf{v}}{\partial t} + \rho (\mathbf{v} \cdot \nabla) \mathbf{v}$$

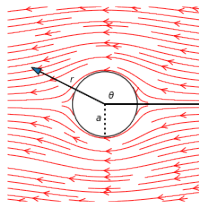
■ Reynolds number

$$\mathcal{R} = \frac{aV\rho}{\eta}$$

■ When $\mathcal{R} \ll 1$

$$-\nabla p + \eta \nabla^2 \mathbf{v} = 0$$

■ Linear relationships between velocity and force at Low Reynolds numbers



Viscous flow around a sphere:

$$v_r = -V \cos \theta \left(1 - \frac{3a}{2r} + \frac{a^3}{2r^3} \right)$$

$$v_\theta = V \sin \theta \left(1 - \frac{3a}{4r} - \frac{a^3}{4r^3} \right)$$

Force on a sphere with velocity $\mathbf{V}$:

$$\mathbf{F} = 6\pi\eta a \mathbf{V}$$

EINSTEIN-SMOLUCHOWSKI: $D = k_B T / f_r$

If $\xi = \frac{dx}{dt}$ is the velocity of a particle, the average is given by equipartition:

$$m \langle \xi^2 \rangle = k_B T$$

A particle moving with respect to the fluid at the speed ξ , experiences a viscous resistance equal to $-f_r \xi$ according to Stokes' formula.

The equation of motion in the x -direction, adding a random fluctuating force owing to the surrounding molecules $X(t)$ is:

$$m \frac{d^2 x}{dt^2} = -f_r \frac{dx}{dt} + X(t)$$

$X(t)$ is indifferently positive and negative, and its magnitude is such that it maintains the agitation of the particle which the viscous resistance is constantly dissipating.

EINSTEIN-SMOLUCHOWSKI: $D = k_B T / f_r$ CONT'D

Multiplying the equation of motion by x and rewriting we have:

$$\frac{m}{2} \frac{d^2 x^2}{dt^2} - m \xi^2 = -\frac{f_r}{2} \frac{dx^2}{dt} + Xx$$

Take the average of the equation of motion for a large number of particles:

- $\langle m \xi^2 \rangle = k_B T$
- $\langle Xx \rangle = 0$
- Define $\frac{d\langle x^2 \rangle}{dt} = z$

$$\frac{m}{2} \frac{dz}{dt} + \frac{f_r}{2} z = k_B T \quad (1)$$

The general solution of Equation 1 is:

$$z = \frac{2k_B T}{f_r} + C e^{-\frac{f_r}{m} t} \quad (2)$$

EINSTEIN-SMOLUCHOWSKI: $D = k_B T / f_r$ CONT'D

The second term in Equation 2 has a very small time constant (nanoseconds for typical small particles) and so vanishes:

$$\frac{d \langle x^2 \rangle}{dt} = \frac{2k_B T}{f_r} \quad (3)$$

Hence, for some period of time τ :

$$\langle x^2 \rangle = 2 \frac{k_B T}{f_r} \tau \quad (4)$$

The diffusion coefficient following Langevin's method is thus:

$$D = \frac{k_B T}{f_r} \quad (5)$$

PROBABILITY DISTRIBUTIONS FOR DIFFUSING PARTICLES

DIFFUSION FOLLOWS BINOMIAL STATISTICS



For a one-dimensional random walk, suppose that a particle steps to the right with a probability p and to the left with a probability q .

$$P(k; n, p) = \frac{n!}{k!(n-k)!} p^k q^{n-k} \quad (6)$$

The displacement of the particles in n trials, $x(n)$, is equal to the number of steps to the right minus the number of steps to the left times the step length, δ :

$$x(n) = [k - (n - k)] \delta = (2k - n) \delta \quad (7)$$

Since we know the distribution of k , we know the distribution of x . The two distributions have the same shape.

DIFFUSION FOLLOWS BINOMIAL STATISTICS CONT'D



The mean displacement of the particle is:

$$\langle x(n) \rangle = (2 \langle k \rangle - n) \delta \quad (8)$$

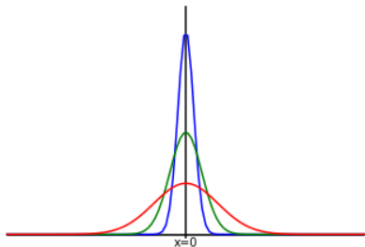
where $\langle k \rangle = np$.

The mean-square displacement is

$$\langle x^2(n) \rangle = \langle [2k - n] \delta \rangle^2 = (4 \langle k^2 \rangle - 4 \langle k \rangle n + n^2) \delta^2 \quad (9)$$

where $\langle k^2 \rangle = (np)^2 + npq$

BINOMIAL TO GAUSSIAN STATISTICS



When n and np are both very large, the binomial distribution, $P(k; n, p)$ is equivalent to:

$$P(k)dk = \frac{1}{(2\pi\sigma^2)^{1/2}} e^{-(k-\mu)^2/2\sigma^2} dk$$

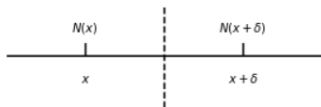
Substitute $x = (2k - n)\delta$, $dx = 2\delta dk$, $p = q = 1/2$, $t = n/\tau$, and $D = \delta^2/2\tau$,

$$P(x)dx = \frac{1}{(4\pi Dt)^{1/2}} e^{-x^2/4Dt} dx$$

- $P(x)dx$ is the probability of finding a particle between x and $x + dx$.
- The variance is $\sigma_x^2 = 2Dt$.
- The standard deviation is $\sigma_x = (2Dt)^{1/2}$.

MACROSCOPIC THEORY

FICK'S FIRST LAW



1. At time t there are $N(x)$ particles at position x , $N(x + \delta)$ particles at $x + \delta$.
2. At time $t + \tau$, half of each set will have stepped left or right.
3. The net number crossing to the right will be: $-\frac{1}{2} [N(x + \delta) - N(x)]$
4. To obtain the net flux, divide by the area normal to the x axis and by τ :

$$J_x = -\frac{1}{2} [N(x + \delta) - N(x)] / A\tau \quad (10)$$

Multiplying Eq. 10 by δ^2/δ^2 and rearrange:

$$J_x = -\frac{\delta^2}{2\tau} \frac{1}{\delta} \left[\frac{N(x + \delta)}{A\delta} - \frac{N(x)}{A\delta} \right]$$

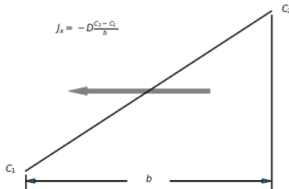
Define $D = \delta^2/2\tau$ to be the diffusion coefficient:

$$J_x = -D \frac{1}{\delta} [C(x + \delta) - C(x)]$$

In the limit $\delta \rightarrow 0$, we get Fick's First Law:

$$J_x = -D \frac{\partial C}{\partial x}$$

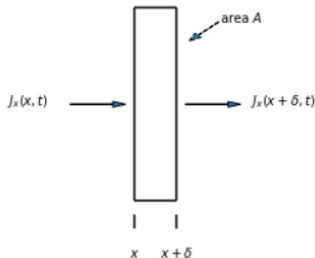
FICK'S FIRST LAW CONT'D



The flux due to a linear concentration gradient $(C_2 - C_1)/b$. Net movement from right to left occurs because there are more particles at the right.

- **Fick's First Law** states that the net flux (at x and t) is proportional to the slope of the concentration function (at x and t). The constant of proportionality is D .
- If the particles are uniformly distributed, the slope is 0, i.e., $\partial C / \partial x = 0$ and $J_x = 0$.
- If the slope is constant, i.e., if $\partial C / \partial x$ is constant, J_x is constant. This occurs when C is a linear function of x .

FICK'S SECOND LAW



Fluxes through the faces of a box. The area of each face is A . The faces are normal to the x axis. In time τ , $J_x(x)A\tau$ particles will enter from the left and $J_x(x + \delta)A\tau$ particles will leave from the right. The volume of the box is $A\delta$.

The number of particles per unit volume in the box must increase at the rate

$$\begin{aligned}\frac{1}{\tau} [C(t + \tau) - C(t)] &= \\ &= -\frac{1}{\tau} [J_x(x + \delta) - J_x(x)] \frac{A\tau}{A\delta} \\ &= -\frac{1}{\delta} [J_x(x + \delta) - J_x(x)]\end{aligned}$$

When $\tau \rightarrow 0$ and $\delta \rightarrow 0$, we get **Fick's Second Law**

$$\frac{\partial C}{\partial t} = -\frac{\partial J_x}{\partial x}$$

THE DIFFUSION EQUATION

Fick's First Law in three dimensions

In three dimensions, we have $J_x = -D\partial C/\partial x$, $J_y = -D\partial C/\partial y$, and $J_z = -D\partial C/\partial z$. These are components of a flux vector:

$$\mathbf{J} = -D\nabla C \quad (11)$$

Fick's Second Law in three dimensions

In three dimensions, we have $\partial C/\partial t = -(\partial J_x/\partial x + \partial J_y/\partial y + \partial J_z/\partial z)$:

$$\frac{\partial C}{\partial t} = -\nabla \cdot \mathbf{J} \quad (12)$$

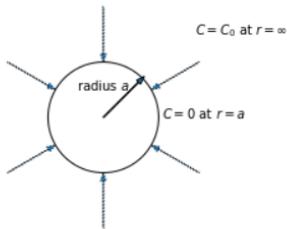
The Diffusion Equation

Combining Eqs. 11 and 12 produces the three-dimensional diffusion equation

$$\frac{\partial C}{\partial t} = D\nabla^2 C \quad (13)$$

DIFFUSION TO CAPTURE

THE SPHERICAL ADSORBER



A spherical adsorber of radius a in an infinite medium containing particles at concentration C_0 . The dashed arrows are lines of flux.

If the problem is spherically symmetric, the flux is radial,

$$J_r = -D \partial C / \partial r$$

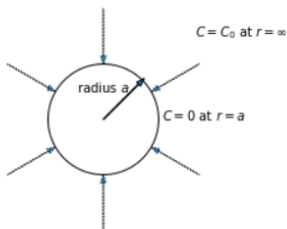
and

$$\frac{\partial C}{\partial t} = D \frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial C}{\partial r} \right)$$

The concentration at $r = a$ is 0. The concentration at $r = \infty$ is C_0 . With these boundary conditions, the diffusion equation has the solution:

$$C(r) = C_0 \left(1 - \frac{a}{r} \right)$$

THE SPHERICAL ADSORBER, CONT'D



A spherical adsorber of radius a in an infinite medium containing particles at concentration C_0 . The dashed arrows are lines of flux.

The flux is

$$J_r = -DC_0 \frac{a}{r^2}$$

The net migration of molecules is radially inward.

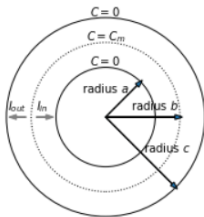
The particles are adsorbed by the sphere at a rate equal to the area, $4\pi a^2$ times the inward flux $-J_r(a)$:

$$I = 4\pi DaC_0$$

This adsorption rate, I , is the diffusion current.

PROBABILITY OF CAPTURE

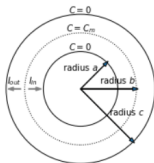
Suppose a particle is released near a spherical adsorber of radius a at a point $r = b > a$? What is the probability that the particle will be adsorbed at $r = a$ rather than wander away for good?



- A spherical shell source, radius b , between a spherical adsorber of radius a and a spherical shell adsorber of radius c .
- Particles released at $r = b$ move inward and are adsorbed at $r = a$ at rate l_{in} or move outward and are adsorbed at $r = c$ at rate l_{out} .
- Their steady-state concentration rises from 0 at $r = a$ to C_m at $r = b$ and then falls again to 0 at $r = c$.

PROBABILITY OF CAPTURE, CONT'D

A boundary value problem



Steady-state diffusion equation

$$\frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial C}{\partial r} \right) = 0$$

Boundary conditions

$$C(r = a) = 0$$

$$C(r = b) = C_m$$

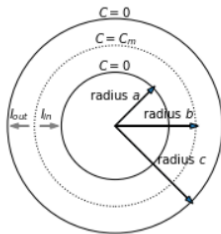
$$C(r = c) = 0$$

The solution

$$C(r) = \begin{cases} \frac{C_m}{1 - a/b} \left(1 - \frac{a}{r} \right) & \text{if } a \leq r \leq b, \\ \frac{C_m}{c/b - 1} \left(\frac{c}{r} - 1 \right) & \text{if } b \leq r \leq c \end{cases}$$

$$J_r(r) = \begin{cases} \frac{-DC_m}{1 - a/b} \frac{a}{r^2} & \text{if } a \leq r \leq b, \\ \frac{DC_m}{c/b - 1} \frac{c}{r^2} & \text{if } b \leq r \leq c \end{cases}$$

PROBABILITY OF CAPTURE, CONT'D CONT'D



- The diffusion current from the spherical shell source to the inner adsorber is

$$I_{in} = 4\pi DC_m \frac{a}{1 - a/b} \quad (14)$$

- The diffusion current from the spherical shell source to the outer adsorber is

$$I_{out} = 4\pi DC_m \frac{c}{c/b - 1} \quad (15)$$

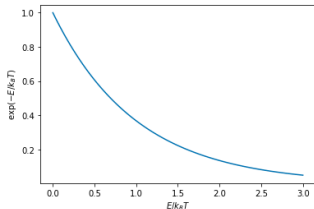
- The probability that a particle released at $r = b$ will be adsorbed at $r = a$ is the ratio:

$$\frac{I_{in}}{I_{in} + I_{out}} = \frac{a(c - b)}{b(c - a)} \quad (16)$$

- In the limit $c \rightarrow \infty$, the probability of being adsorbed is a/b .

EINSTEIN RELATION

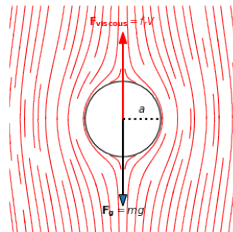
SEDIMENTATION OF SMALL PARTICLES



The density of particles suspended in a fluid in a gravitational field is given by the Boltzmann distribution:

$$\rho(z) = \rho_0 e^{-mgz/k_B T}$$

because $E = mgz$



Each particle sediments at a velocity given by the balance between gravitational force ($\mathbf{F} = m\mathbf{g}$) and viscous force ($\mathbf{F} = -\zeta \mathbf{V}$).

EINSTEIN RELATION

Upward flux due to diffusion is given by Fick's First Law

$$J_z^{diff} = D \times \frac{mg}{k_B T} \times \rho_0 e^{-\frac{mgz}{k_B T}}$$

Downward flux due to sedimentation is given by the product of velocity and particle density

$$J_z^{sed} = -\frac{mg}{f_r} \times \rho_0 e^{-\frac{mgz}{k_B T}}$$

At steady state, $J_z^{diff} + J_z^{sed} = 0$:

$$D \times \frac{mg}{k_B T} \times \rho_0 e^{-\frac{mgz}{k_B T}} - \frac{mg}{f_r} \times \rho_0 e^{-\frac{mgz}{k_B T}} = 0$$
$$\frac{D}{k_B T} - \frac{1}{f_r} = 0$$

Hence,

$$D = \frac{k_B T}{f_r}$$