COUNTING MOLECULES WITH RECEPTORS

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Week 6

COUNTING MOLECULES BY CAPTURE

SOLVING THE STEADY-STATE DIFFUSION EQUATION



- Consider a spherical cell of radius a immersed in an unbounded medium.
- The medium contains in low concentration some molecules with diffusion constant *D*.
- The local concentration, $c(\mathbf{x}, t)$ will be expressed in molecules per unit volume.

The spatial and temporal variation of $c(\mathbf{x}, t)$ is governed by the diffusion equation

$$D\nabla^2 c = \frac{\partial c}{\partial t} \tag{1}$$

In steady state, the current *I* of molecules to the cell, in molecules per second, is given by the product of surface area $4\pi a^2$ and the flux, *J*, from Fick's First Law:

$$I = 4\pi a D c_{\infty}$$
 (2)

where c_{∞} is the concentration far from the cell.

ELECTROSTATICS ANALOGY

Electrostatics

Laplace's Equation for charge-free space

 $\nabla^2 \phi = 0$

Electric field vector

 $\mathbf{E} = \mathbf{\nabla}\phi$

Total electric charge Q on a surface:

$$Q = \frac{1}{4\pi} \int_{\mathsf{S}} \mathbf{E} \cdot d\mathbf{s}$$

Diffusion

Time-independent diffusion equation:

 $\nabla^2 \mathbf{C} = \mathbf{O}$

Diffusive current density:

 $\mathbf{J} = -D\boldsymbol{\nabla}\mathbf{c}$

Total diffusive current entering closed surface:

$$I = \int_{S} \mathbf{J} \cdot d\mathbf{s}$$

 $I=4\pi CDc_{\infty}$

where C is the electrical capacitance of an object in cgs units

COUNTING MOLECULES USING A PATCHY SPHERE





- The path of a diffusing molecule that has touched the surface of a cell of radius *a* at a sequence of points A, B, . . . F.
- The receptor patches, shown shaded, are of radius s.

The probability that a molecule at r = s + a from the sphere of radius *a* will hit the surface before escaping:

$$P_{\rm s}=rac{a}{a+s}$$

The probability that a molecule at r = a + s executes ntrips to the surface, separated by reappearances at r = a + s before escaping:

 $P_s^n(1-P_s)$

The average number of surface trips is:

$$\bar{n} = \sum_{n=0}^{\infty} n P_s^n (1 - P_s)$$
$$= \frac{P_s}{1 - P_s}$$
$$= \frac{a}{s}$$

PATCHY RECEPTOR, CONT'D



• The probability of not hitting a receptor patch in a single random encounter is $\beta = 1 - (Ns^2/4a^2)$.

• If each contact is an independent attempt, the probability that a molecule starting at r = a + s survives all contacts until it escapes to infinity is:

$$P_{esc} = \sum_{n=0}^{\infty} \beta^n n P_s^n (1 - P_s) = \frac{1 - P_s}{1 - \beta P_s} = \frac{4a}{4a + N_s}$$

 \circ Since 1 – P_{esc} is the fraction of all arriving molecules that ultimately are captured, the resulting current is:

$$\frac{J}{J_{max}} = \frac{Ns}{4a + Ns}$$

MEASURING CONCENTRATION WITH RECEPTORS

MEAN RECEPTOR OCCUPANCY



- Time-averaged occupancy of a receptor is given by a single dissociation constant, *K*
- In equilibrium at concentration *c*, the expected average occupancy is:

$$\bar{p} = \frac{c}{c + c_{1/2}}$$

Law of Mass Action

$$\bar{p} = \frac{[R \cdot L]}{[R][L] + [R \cdot L]}$$

$$= \frac{[L]}{[L] + k_{off}/k_{on}}$$

AN ENTROPIC VIEW OF RECEPTOR OCCUPANCY



Physics of the unbound state

• Energy = $L\epsilon_{sol}$

• Multiplicity =
$$\frac{\Omega!}{L!(\Omega-L)!} \approx \frac{\Omega^L}{L!}$$

• Weight =
$$\frac{\Omega^L}{L!} e^{-\beta L \epsilon_{sol}}$$



Physics of the bound state

• Energy = $(L - 1)\epsilon_{sol} + \epsilon_b$

• Multiplicity
$$= \frac{\Omega!}{(L-1)!(\Omega-L+1)!} \approx \frac{\Omega^{L-1}}{(L-1)!}$$

• Weight =
$$\frac{\Omega^{L-1}}{(L-1)!} e^{-\beta[(L-1)\epsilon_{sol}+\epsilon_b]}$$

The probability of binding is the weight of the bound states divided by the sum of the weights of the bound and unbound states:

$$\bar{p} = \frac{\frac{\Omega^{L-1}}{(L-1)!} e^{-\beta[(L-1)\epsilon_{sol} + \epsilon_b]}}{\frac{\Omega^{L-1}}{(L-1)!} e^{-\beta[(L-1)\epsilon_{sol} + \epsilon_b]} + \frac{\Omega^L}{L!} e^{-\beta L\epsilon_{sol}}}$$

Multiply top and bottom by $(L!/\Omega^L)e^{\beta L\epsilon_{sol}}$:

$$\bar{p} = \frac{(L/\Omega)e^{-\beta\Delta\epsilon}}{1+(L/\Omega)e^{-\beta\Delta\epsilon}}$$

where $\Delta \epsilon = \epsilon_b - \epsilon_{sol}$.

The overall volume of the box is V_{box} , so we can write ligand concentration as $c = L/\Omega V_{box}$. This results in:

$$\bar{p} = rac{(c/c_0)e^{-eta\Delta\epsilon}}{1+(c/c_0)e^{-eta\Delta\epsilon}}$$

where $c_0 = 1/V_{box}$.

CALCULATING CONCENTRATION FROM RECEPTOR OCCUPANCY

TEMPORAL DYNAMICS OF RECEPTOR OCCUPANCY





Since
$$\bar{p} = 1/2$$
 when $c = c_{1/2}$
 $\tau_b = (4Dsc_{1/2})^{-1}$



Fractional error in measuring concentration is related to mean and variance in receptor occupancy:

$$\frac{\delta c}{c} = \frac{\delta \bar{p}}{\bar{p}(1-\bar{p})}$$

If the only information about c is the function p(t) for one receptor recorded for a time T, the cell can extract this information by forming its average:

$$p_T = \frac{1}{T} \int_{t_1}^{t_1+T} p(t) dt$$

If p_T is taken as the \bar{p} , it can be used to derive *c*:

$$c/c_{1/2} = p_T/(1-p_T)$$

Variance in measurements of \bar{p} can be computed from the autocorrelation function of p(t)

$$G(\tau) = \langle p(t)p(t+\tau) \rangle$$



$$p_T = \frac{1}{T} \int_{t_1}^{t_1+T} p(t) dt$$

• A useful tool is the autocorrelation function of *p*(*t*):

$$\mathsf{G}(au) = \langle \mathsf{p}(t)\mathsf{p}(t+ au)
angle$$

- $\circ \ \mathsf{G}(au)$ is an even function of au: $\mathsf{G}(au) = \mathsf{G}(- au)$
- From the definition of p_T

$$p_{T}^{2} = \frac{1}{T^{2}} \int_{t_{1}}^{t_{1}+T} dt' \int_{t_{1}}^{t_{1}+T} p(t)p(t')dt$$

Introducing the autocorrelation function, we obtain:

$$\left< p_T^2 \right> = rac{1}{T^2} \int_0^T dt' \int_0^T G(t'-t) dt$$

- $\circ~$ Consider a large number n pairs of observations, one at t the other at t + $\tau.$
- Segregate the pairs in which the first observation is bound, i.e., p(t) = 1.
 - ▷ If *n* is large, there will be about $n\bar{p}$ pairs
 - \triangleright The number with $p(t + \tau) = 1$ will be $nG(\tau)$
 - ▷ These "1,1" pairs are the only ones for which $p(t)p(t + \tau) \neq 0$
- $\circ~$ Shift the time of the second observation from $t+\tau$ to $t+\tau+d\tau$
 - Some of the "1,1" pairs will become "1,0" pairs. This number is

$nGd au/ au_b$

Some of the "1,0" pairs will become "1,1" pairs, this number is

 $n(\bar{p}-G)[\bar{p}/(1-\bar{p})]d\tau/\tau_b$

Fractional error in measuring concentration is from the counting statistics of the total number of measured molecules! • We should now have $nG(\tau + d\tau)$ "1,1" pairs, which requires:

$$dG = -Gd au/ au_b + (ar{p} - G)[ar{p}/(1 - ar{p})]d au/ au_b$$

• Integrating and requiring that
$$G(o)=ar{p}$$

$$G(\tau) = \bar{p}^2 + \bar{p}(1-\bar{p}) \exp\left[-\frac{|\tau|}{(1-\bar{p})\tau_b}\right]$$

• Assuming $T \gg \tau_b$:

$$\left\langle p_{T}^{2} \right\rangle - \left\langle p_{T} \right\rangle^{2} = rac{2\bar{p}(1-\bar{p})^{2}\tau_{b}}{T}$$

• For the rms error in \bar{c} we get

$$\frac{\Delta c_{rms}}{\bar{c}} = \sqrt{\frac{2\tau_b}{T\bar{p}}} = \sqrt{\frac{2}{\nu}}$$

where
$$\nu = 4Ds\overline{c}(1-\overline{p})T$$